

# Influence of microstructure on strength and fracture toughness of $\beta$ -Sialon

K. KISHI, S. UMEBAYASHI, E. TANI

Government Industrial Research Institute, Kyushu Shuku-machi, Tosu-shi, Saga 841, Japan

$\beta$ -Sialon ( $\text{Si}_{6-z}\text{Al}_z\text{O}_z\text{N}_{8-z}$ ) with  $z = 0.5$  was fabricated by hot-pressing of a spray dried mixture of  $\alpha$ - $\text{Si}_3\text{N}_4$  and aluminium-isopropoxide solution. Phase composition, flexural strength and microstructure of a sintered body were investigated. Phases identified by XRD were  $\beta$ -Sialon and a small amount of O'-Sialon. The flexural strength (three-point bending) was about  $1500 \text{ MN m}^{-2}$ . This value, about three times higher than that of  $\beta$ -Sialon fabricated from  $\alpha$ - $\text{Si}_3\text{N}_4$  and  $\alpha$ - $\text{Al}_2\text{O}_3$  powder, was mainly due to the homogeneous microstructure without large defects such like clusters of large grains.  $\beta$ -Sialon was heat treated at  $2000^\circ \text{C}$  for 2 h in 4 MPa  $\text{N}_2$  to develop elongated  $\beta$ -Sialon grains with high aspect ratio. Microstructure, flexural strength and fracture toughness ( $K_{\text{Ic}}$ ) of it were investigated. Both strength and  $K_{\text{Ic}}$  were lower than those of hot-pressed sample, even though an elongated microstructure was achieved. This fact showed that the toughening of  $\beta$ -Sialon with elongated grains could not be achieved without grain boundary phase which resulted in a crack deflection.

## 1. Introduction

$\beta$ -Sialon ( $\text{Si}_{6-z}\text{Al}_z\text{O}_z\text{N}_{8-z}$ ) is of interest for applications in ceramic components for heat engine and other high temperature engineering [1], because of their excellent properties such like high temperature strength, high oxidation and corrosion resistance.

Commonly,  $\beta$ -Sialon is fabricated from the powders mixture of  $\text{Si}_3\text{N}_4$ ,  $\text{Al}_2\text{O}_3$  and  $\text{AlN}$  [2, 3] or  $\text{Si}_3\text{N}_4$ ,  $\text{SiO}_2$  and  $\text{AlN}$  [4, 5]. However the strength of these  $\beta$ -Sialon [6-8] is still low compared with  $\text{Si}_3\text{N}_4$  based ceramics such as  $\text{Si}_3\text{N}_4$  with  $\text{Al}_2\text{O}_3$  and  $\text{Y}_2\text{O}_3$  [9] or  $\text{MgO}$  [10]. The strength of  $\beta$ -Sialon is largely influenced by the existence of unhomogeneous parts such as clusters of large grains or porous parts because these unhomogeneous parts become a fracture origin [6, 7]. The formation of these unhomogeneous parts assumed to originate in the unhomogeneous mixing and dispersing of starting powders. Thus in the parts where much liquid phase formed during sintering and grain growth resulted and in the opposite sense porous parts were formed. It is considered that the strength of  $\beta$ -Sialon must be improved if these unhomogeneous parts as a fracture origin can be removed.

For this purpose, a aluminium alkoxide used as a solution was an effective starting material for  $\beta$ -Sialon to get homogeneous microstructure because of its uniformity and reaction as the surface modified of  $\text{Si}_3\text{N}_4$  powder [11-13], and the strength of  $\beta$ -Sialon fabricated using aluminium alkoxide instead of  $\text{Al}_2\text{O}_3$  powder must be greatly improved.

It was believed that the high strength and the high fracture toughness of  $\text{Si}_3\text{N}_4$  with  $\text{Y}_2\text{O}_3$  and  $\text{Al}_2\text{O}_3$  originated in their elongated  $\beta$ - $\text{Si}_3\text{N}_4$  grains with high aspect ratio (length/width) [14, 15], and the low fracture toughness of  $\beta$ -Sialon originated in the equiaxed

$\beta$ -Sialon grains from the standpoint of crack deflection. The grain morphology of  $\beta$ -Sialon can be controlled from equiaxed to elongated one by heating it at high temperature under applied  $\text{N}_2$  gas atmosphere [16].

To obtain a  $\beta$ -Sialon with a high strength and high fracture toughness,  $\beta$ -Sialon with  $z = 0.5$  was fabricated by hot-pressing a spray dried mixture of  $\alpha$ - $\text{Si}_3\text{N}_4$  and aluminium-isopropoxide solution, and heat treating at  $2000^\circ \text{C}$  for 2 h under 4 MPa  $\text{N}_2$  gas atmosphere to develop elongated  $\beta$ -Sialon grains. Some mechanical properties were measured in comparison with those of  $\beta$ -Sialon fabricated from a powder mixture of  $\alpha$ - $\text{Si}_3\text{N}_4$  and  $\alpha$ - $\text{Al}_2\text{O}_3$ . This paper describes the strength and the fracture toughness of  $\beta$ -Sialon and discusses the influence of microstructure on them.

## 2. Experimental techniques

$\alpha$ - $\text{Si}_3\text{N}_4$  powder was purchased from Toyo Soda Co. Ltd (TS-7) and aluminium-isopropoxide ( $\text{Al}(\text{Oi-Pr})_3$ , extra pure reagent) from Nakarai Chemicals Co. Ltd.  $\text{Al}(\text{Oi-Pr})_3$  was dissolved in dry  $n\text{-C}_6\text{H}_{14}$  and the solution was filtered to remove the aluminium hydroxide on the surface of  $\text{Al}(\text{Oi-Pr})_3$ . The precipitant was fired and the amount of  $\text{Al}(\text{Oi-Pr})_3$  in the filtrate was determined.  $\alpha$ - $\text{Si}_3\text{N}_4$  powder was added to the filtrate in the ratio of  $\text{Al}_2\text{O}_3 : \text{Si}_3\text{N}_4 = 9.01 : 90.99$ , where the Al/Si ratio was the same as  $z = 0.5$  in the formula of  $\text{Si}_{6-z}\text{Al}_z\text{O}_z\text{N}_{8-z}$ . The mixture of  $\text{Si}_3\text{N}_4$  and  $\text{Al}(\text{Oi-Pr})_3/\text{C}_6\text{H}_{14}$  was ball milled for 20 h with  $\text{Si}_3\text{N}_4$  balls and spray dried. The spray dried powder was calcined at  $300^\circ \text{C}$  for 2 h in air. The calcined powder was hot-pressed at  $1850^\circ \text{C}$  for 1 h under 29 MPa in 1 atm  $\text{N}_2$  using a 30 mm  $\times$  30 mm carbon mould. A hot-pressed sample was prepared from the powders mixture of

TABLE I Densities and phases of hot-pressed samples

Starting material	Bulk density (g cm <sup>-3</sup> )	Phases
Al(Oi-Pr) <sub>3</sub>	3.14	$\beta$ + O'(w)
Al <sub>2</sub> O <sub>3</sub>	3.14	$\beta$ + $\alpha$ (tr)

$\beta$ :  $\beta$ -Sialon.

$\alpha$ :  $\alpha$ -Si<sub>3</sub>N<sub>4</sub>.

O': Si<sub>2</sub>N<sub>2</sub>O(s.s.).

$\alpha$ -Al<sub>2</sub>O<sub>3</sub> (Iwatani Chemicals Co. Ltd RA-40) and  $\alpha$ -Si<sub>3</sub>N<sub>4</sub> for comparison.

A hot-pressed sample using Al(Oi-Pr)<sub>3</sub> was heat treated at 2000°C in 4 MPa N<sub>2</sub> for 2 h.

The density of as hot-pressed and heat treated samples was determined by water immersion method. The phase composition was determined by powder X-ray diffraction analysis (XRD). The polished and fractured surfaces were observed with optical and scanning electron microscopes (SEM).

The specimens (3 mm × 3 mm × 30 mm) for flexural strength measurement were prepared by grinding with No. 600 diamond wheel and chamfering with No. 800 SiC abrasive paper, and then they were fired at 1200°C for 2 h in air to reduce the influence of surface machining flaws. Flexural strength ( $\sigma_f$ ) measurement was performed at room temperature using a three-point loading device with a span of 20 mm and a crosshead speed of 0.5 mm min<sup>-1</sup>. The fracture toughness ( $K_{Ic}$ ) was measured by Vickers indentation method with load of 10 kg by using the equation given by Niihara [17].

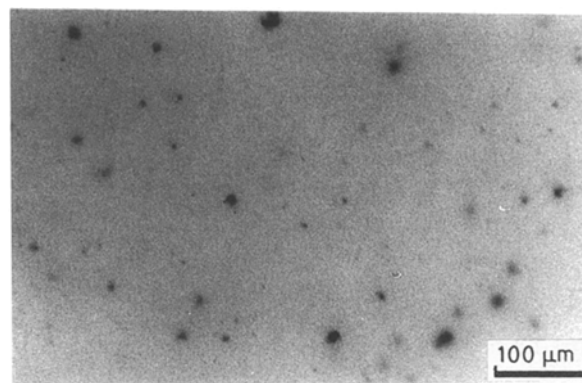


Figure 1 Optical micrograph of sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>.

### 3. Results and discussion

#### 3.1. As hot-pressed sample

Table I showed the bulk density and phase composition of hot-pressed samples fabricated from Si<sub>3</sub>N<sub>4</sub>-Al(Oi-Pr)<sub>3</sub> and Si<sub>3</sub>N<sub>4</sub>-Al<sub>2</sub>O<sub>3</sub>. The density of both samples was 3.14 g cm<sup>-3</sup> (> 99% theoretical). Sample using Al(Oi-Pr)<sub>3</sub> consisted of  $\beta$ -Sialon and a small amount of O'-Sialon. This phase composition corresponded to the phase diagram of Si-Al-O-N system given by Jack [1] or Gauckler *et al.* [18]. On the other hand,  $\alpha$ -Si<sub>3</sub>N<sub>4</sub> remained and no O'-Sialon was observed in sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>. This could be explained by the unhomogeneous dispersion of aluminium and oxygen, in the sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> powder. That is Al<sub>2</sub>O<sub>3</sub>-rich parts were formed where more glassy phase was formed and Al<sub>2</sub>O<sub>3</sub> lacking parts were  $\alpha$ -Si<sub>3</sub>N<sub>4</sub> remained.

Fig. 1 showed an optical micrograph of sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>. There could be found a lot of unhomogeneous parts observed as black spots. These black spots included large amounts of glassy phase and were easily chemically etched with HF. No such large black spots could be found in the sample using Al(Oi-Pr)<sub>3</sub>. Fig. 2 shows SEM photographs of the samples. In the sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>, many pores were formed by chemical etching (Fig. 2a) and clusters of large grains (Fig. 2b) were observed. Sample using Al(Oi-Pr)<sub>3</sub> showed a homogeneous microstructure consisting of

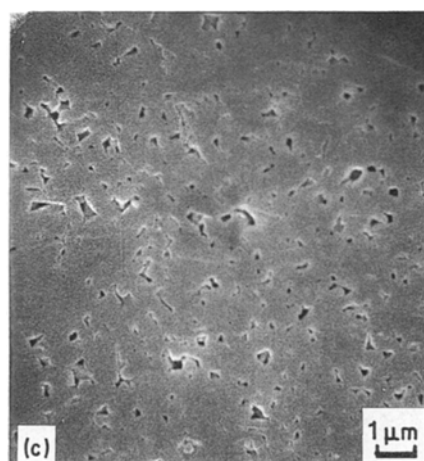
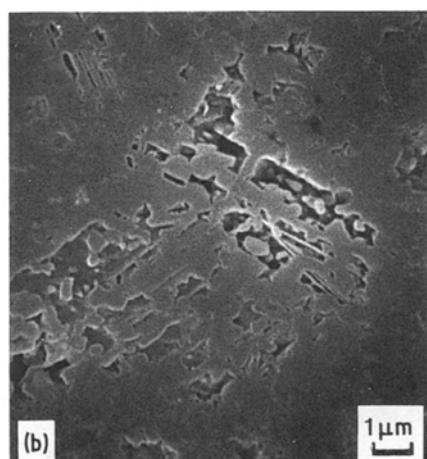
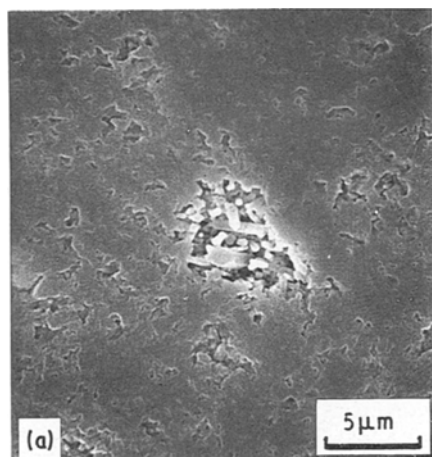


Figure 2 SEM photographs of samples (polished and etched with HF and HNO<sub>3</sub>): (a), (b) sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>, (c) sample using Al(Oi-Pr)<sub>3</sub>.

TABLE II Mechanical properties of hot-pressed samples

Starting material	$\sigma_f$ (MN m <sup>-2</sup> )	$Hv$ (GN m <sup>-2</sup> )	$K_{Ic}$ (MN m <sup>-3/2</sup> )
Al(Oi-Pr) <sub>3</sub>	1480	16.0	3.3
Al <sub>2</sub> O <sub>3</sub>	484	15.7	3.2

0.5–1  $\mu\text{m}$   $\beta$ -Sialon grains and a small amount of grain boundary phase removed by chemical etching (Fig. 2c). The distribution of the grain boundary phase was more homogeneous compared with the sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>.

These results suggested that the use of Al(Oi-Pr)<sub>3</sub> solution was more desirable to get  $\beta$ -Sialon with homogeneous microstructure.

Table II showed the mechanical properties of hot-pressed samples. Vickers hardness and  $K_{Ic}$  of both  $\beta$ -Sialons showed approximately the same value of 16 GN m<sup>-2</sup> and 3.3 MN m<sup>-3/2</sup>, respectively. However, strength of the sample using Al(Oi-Pr)<sub>3</sub> was about 1500 MN m<sup>-2</sup>, about three times higher than that of the sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> and conventional  $\beta$ -Sialon [6–8].

Fig. 3 showed the fracture origins of the samples. The fracture origins in the sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (Fig. 3a) were large internal defects originated in unhomogeneous parts shown in Figs 1 and 2. The fracture origins of sample using Al(Oi-Pr)<sub>3</sub> were internal defects (Fig. 3b) and machining flaws (Fig. 3c). However, the size of the internal defects were much smaller than those in the sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>. Fig. 4 showed SEM photographs of a typical internal defect as a fracture origin in the sample formed from Al(Oi-Pr)<sub>3</sub>. These defects consisted of a slightly larger size of grains compared with the matrix. And the shape of defects was flat and lying perpendicular to hot-pressing direction. These internal defects resulted from the Al<sub>2</sub>O<sub>3</sub> concentrated parts and the size of these Al<sub>2</sub>O<sub>3</sub> concentrated parts (about 10  $\mu\text{m}$ ) was assumed to

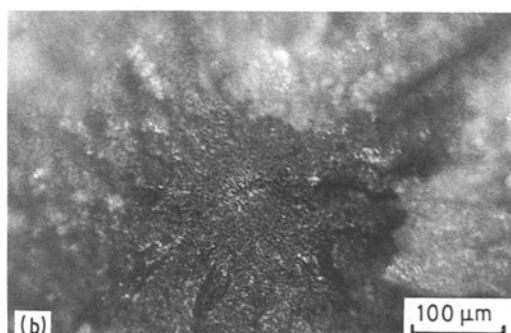
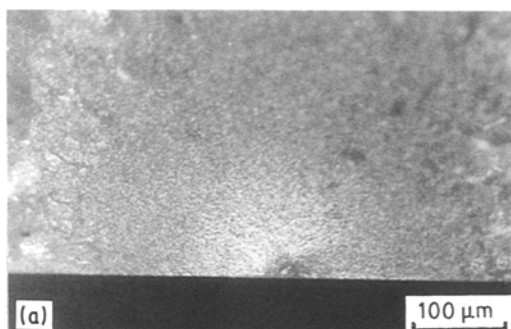


TABLE III Mechanical properties of heat-treated sample

$\sigma_f$ (MN m <sup>-2</sup> )	$Hv$ (GN m <sup>-2</sup> )	$K_{Ic}$ (MN m <sup>-3/2</sup> )
956	13.9	2.8

have been limited by the size of the particles formed by spray drying.

Thus, the higher strength of the sample using Al(Oi-Pr)<sub>3</sub> was mainly due to the homogeneous microstructure without large defects. And it was expected that if the influence of surface machining flaws could be removed completely, the strength must be higher as  $\beta$ -Sialon was very sensitive to microcracks for its low  $K_{Ic}$ , and the lowest strength of it would be controlled by the maximum size of spray dried particles.

Fig. 5 shows a fractured specimen after the bending test. The cracks formed symmetrical fragments from a fracture origin and the shape of these fragments were almost the same as the distribution of stress before fracture [19]. This result suggests that only one defect where a crack began to move became a fracture origin and propagation of this crack released the stress of loading instantaneously.

### 3.2. Heat-treated sample

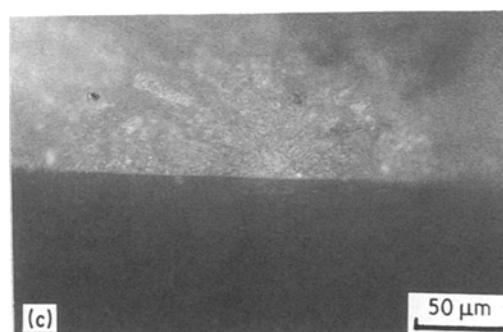
The suppressed thermal decomposition of  $\beta$ -Sialon and the weight loss and the change of density after heat treatment at 2000°C for 2 h in 4 MPa N<sub>2</sub> atmosphere were not observed. The crystalline phase identified by XRD in the heat treated sample was  $\beta$ -Sialon.

Table III shows the mechanical properties of the heat treated sample. Both hardness and  $K_{Ic}$  decreased to the value of 13.9 GN m<sup>-2</sup> and 2.8 MN m<sup>-3/2</sup>, respectively. Room temperature strength also decreased to about 950 MN m<sup>-2</sup>.

The microstructure of heat treated sample (Fig. 6) consisted of grown and elongated  $\beta$ -Sialon grains, and a small amount of grain boundary phase. This microstructure was almost the same as that of the Si<sub>3</sub>N<sub>4</sub> sintered body with high fracture toughness [20].

In general, it was believed that the lower fracture toughness of pure  $\beta$ -Sialon was due to its microstructure with equiaxed grains where crack deflection was difficult. However, according to this study, this is not correct at least in the case of pure  $\beta$ -Sialon. In the case of MgO-doped [21] or Y<sub>2</sub>O<sub>3</sub>-doped  $\beta$ -Sialon [22], the

Figure 3 Optical micrographs of fracture origins: (a) defect in sample using  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> ( $\sigma_f = 447$  MN m<sup>-2</sup>), (b) defect in sample using Al(Oi-Pr)<sub>3</sub>, ( $\sigma_f = 1379$  MN m<sup>-2</sup>), (c) surface machining flaw in sample using Al(Oi-Pr)<sub>3</sub> ( $\sigma_f = 1555$  MN m<sup>-2</sup>).



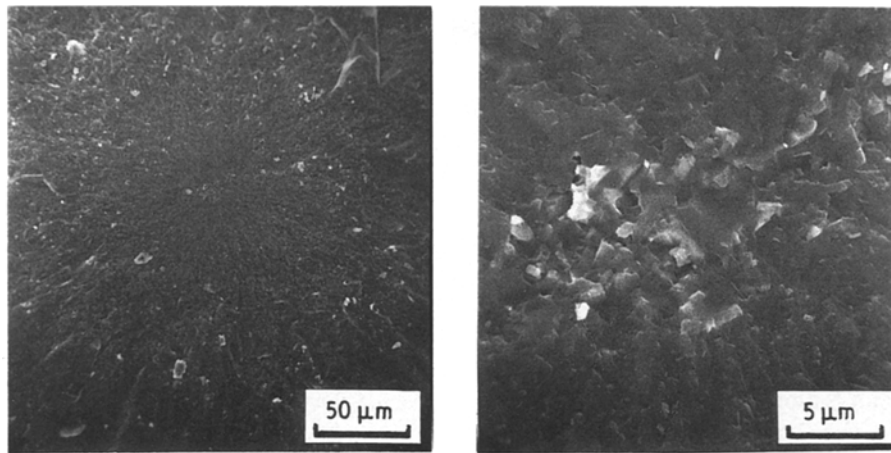


Figure 4 SEM photographs of a defect as a fracture origin (same specimen as Fig. 3b).

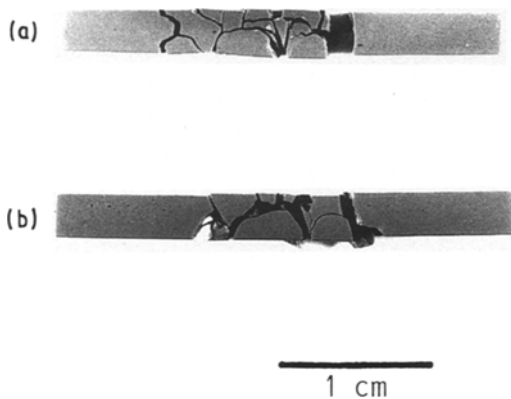


Figure 5 Fractured specimen of sample using  $\text{Al}(\text{O}i\text{-Pr})_3$  after bending test. (a)  $\sigma_f = 1402 \text{ MN m}^{-2}$ ; (b)  $\sigma_f = 1399 \text{ MN m}^{-2}$ .

concentration of Mg and O or Y and O in the grain boundary phase and the formation of a low melting glassy phase promoted the development of elongated grains. This grain boundary phase was weak compared with  $\beta$ -Sialon grains. Therefore, crack deflection took place along the elongated  $\beta$ -Sialon grains and resulted in the improvement of fracture toughness. In the case of pure  $\beta$ -Sialon, it was considered that according to the presence of a small amount of grain boundary phase and a small difference in the physical properties between the grain boundary phase

and  $\beta$ -Sialon grains, no crack deflection took place along the elongated grains. In addition to the increase in defect size by grain growth, the presence of elongated  $\beta$ -Sialon grains might be one reason for the decrease in the strength and  $K_{Ic}$  of  $\beta$ -Sialon.

The fracture surface of heat treated sample was mostly occupied by transgranular fracture except for the long grain boundary lying parallel to a crack propagation (Fig. 7). This is assumed to indicate that the existence of this kind of long grain boundary makes the crack propagation easy and the crack deflection that was observed in  $\text{Si}_3\text{N}_4$  [23] does not occur. In the case of this sample, this was the reason for the degradation of the strength and  $K_{Ic}$ .

From these results, it can be concluded that the toughening of  $\beta$ -Sialon with elongated microstructure could not be achieved without the existence of a weak grain boundary phase where a crack could easily propagate. If there was no weak grain boundary phase for crack deflection, the existence of elongated grains results in a decrease in both toughness and strength.

#### 4. Conclusion

$\beta$ -Sialon with  $z = 0.5$  was fabricated by hot-pressing of spray dried mixture of  $\alpha\text{-Si}_3\text{N}_4$  and  $\text{Al}(\text{O}i\text{-Pr})_3$  solution. Phase composition, flexural strength and microstructure of a sintered body were investigated.

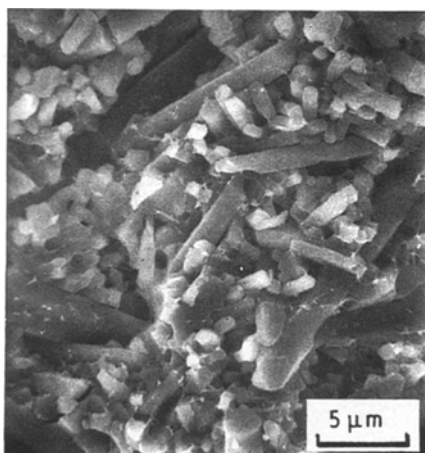


Figure 6 Microstructure of heat-treated sample.

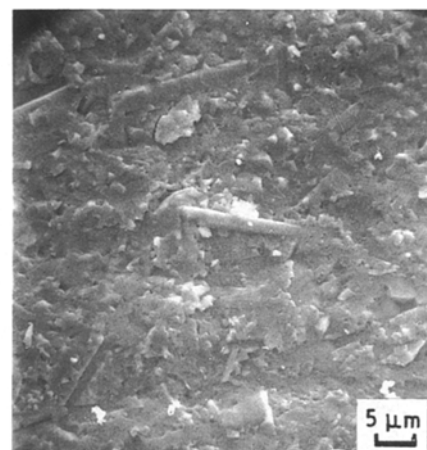


Figure 7 Fracture surface of heat-treated sample.

Phases identified by XRD were  $\beta$ -Sialon and a small amount of O'-Sialon, but no  $\alpha$ -Si<sub>3</sub>N<sub>4</sub> that remained in the hot-pressed sample from the powder mixture of  $\alpha$ -Si<sub>3</sub>N<sub>4</sub> and  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> was observed. This assumed to show that starting powder using Al(Oi-Pr)<sub>3</sub> was more homogeneous than that using Al<sub>2</sub>O<sub>3</sub> powder. The flexural strength (three-point bending) was about 1500 MN m<sup>-2</sup>. This value was remarkably higher than that of the sample using Al<sub>2</sub>O<sub>3</sub> powder mainly due to the homogeneous microstructure without large internal defects such as clusters of large grains which were observed in sample using Al<sub>2</sub>O<sub>3</sub> powder.

The hot-pressed  $\beta$ -Sialon using Al(Oi-Pr)<sub>3</sub> was heat treated at 2000°C for 2 h in 4 MPa N<sub>2</sub> to form an elongated microstructure with high aspect ratio  $\beta$ -Sialon grains. Flexural strength, fracture toughness ( $K_{Ic}$ ) and microstructure were investigated.

Both the strength and  $K_{Ic}$  were decreased compared with the as hot-pressed sample even though the elongated microstructure with high aspect ratio  $\beta$ -Sialon grains was achieved. And a crack deflection which was observed in Si<sub>3</sub>N<sub>4</sub>- and Y<sub>2</sub>O<sub>3</sub>-doped  $\beta$ -Sialon was not observed. These results are assumed to be due to the existence of elongated  $\beta$ -Sialon grains without a weak grain boundary phase to result crack deflections.

## References

1. K. H. JACK, *J. Mater. Sci.* **11** (1976) 1135.
2. M. MITOMO *et al.*, *Yogyo-kyokai-shi* **86** (1978) 526.

3. R. R. WILLS, R. W. STEWART and J. M. WIMMER, *J. Amer. Ceram. Soc.* **60** (1977) 64.
4. M. MITOMO and N. KURAMOTO, *Yogyo-kyokai-shi* **87** (1979) 526.
5. M. H. LEWIS *et al.*, *J. Mater. Sci.* **12** (1977) 61.
6. S. UMEBAYASHI *et al.*, *Yogyo-kyokai-shi* **90** (1982) 615.
7. S. UMEBAYASHI *et al.*, *ibid.* **92** (1984) 35.
8. M. MITOMO *et al.*, *ibid.* **88** (1980) 298.
9. A. TSUGE and K. NISHIDA, *Amer. Ceram. Soc. Bull.* **57** (1978) 424.
10. F. F. LANGE, *J. Amer. Ceram. Soc.* **57** (1974) 84.
11. K. KISHI *et al.*, *Yogyo-kyokai-shi* **96** (1988) 698.
12. K. KISHI *et al.*, *ibid.* **92** (1984) 231.
13. K. KISHI *et al.*, *ibid.* **93** (1985) 629.
14. J. HEINRICH and M. BOHMAN, *Commun. Am. Ceram. Soc.* **67** (1984) C-75.
15. G. WOTTING and G. ZIEGLER, *Ceram. Int.* **10** (1984) 18.
16. K. KISHI, unpublished.
17. K. NIIHARA, *Seramikkusu* **20** (1985) 12.
18. L. J. GAUCKLER, H. L. LUKAS and G. PEZTOW, *J. Amer. Ceram. Soc.* **58** (1975) 346.
19. Y. MATSUNO, M. ITO and H. OKUDA, *Seramikkusu* **16** (1981) 549.
20. E. TANI *et al.*, *J. Mater. Sci. Lett.* **4** (1985) 1454.
21. M. H. LEWIS *et al.*, *J. Mater. Sci.* **12** (1977) 61.
22. M. H. LEWIS *et al.*, *ibid.* **15** (1980) 103.
23. E. TANI *et al.*, *Amer. Ceram. Soc. Bull.* **65** (1986) 1311.

*Received 5 April  
and accepted 4 September 1989*